

## Control of Magnetic Dimensionality Through Application of Pressure in Hydrogen Bonded Magnetic Coordination Polymers

John A. Schlueter<sup>1</sup>, Gregory J. Halder<sup>1</sup>, Karena W. Chapman<sup>2</sup>, and Jamie L. Manson<sup>3</sup>

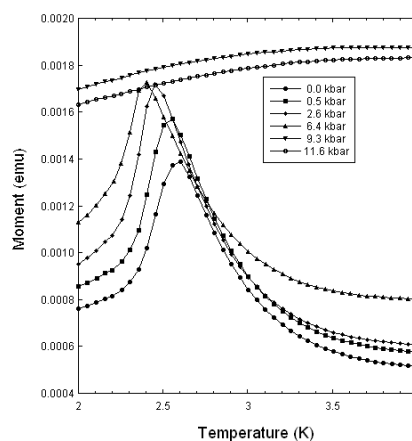
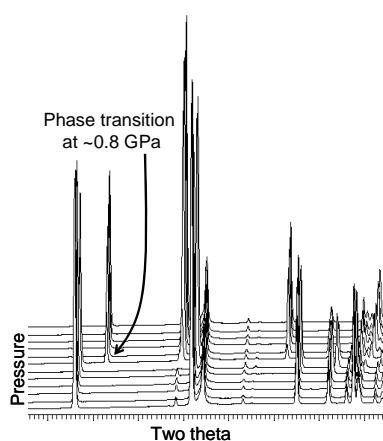
<sup>1</sup>Materials Science Division, Argonne National Laboratory, Argonne, IL, 60439, USA

<sup>2</sup>X-ray Science Division, Advanced Photon Source, Argonne National Lab., Argonne, IL 60439, USA

<sup>3</sup>Department of Chemistry, Eastern Washington University, Cheney, WA, 99004, USA.

Email: JASchlueter@anl.gov

We have studied the use of very strong hydrogen bonds as structure-directing entities in magnetic coordination polymers such as  $[\text{Cu}(\text{pyz})_2(\text{HF}_2)]X$  (pyz = pyrazine;  $X$  = tetrahedral or octahedral anion) [1] and  $\text{CuF}_2(\text{H}_2\text{O})_2(\text{pyz})$  [2]. In fluorinated systems, competition for hydrogen bonding interactions enables the stabilization of polymorphs with small energy differences [3]. This competition also renders pressure an effective tool for inducing structural changes that lead to significant changes in magnetic properties. We report a novel pressure-induced reorientation of the Jahn-Teller axis in the  $\text{CuF}_2(\text{H}_2\text{O})_2(\text{pyz})$  system that results in a dramatic change in the magnetic dimensionality. The incorporation of similar magnetic systems into conducting charge transfer salts will be discussed.



Pressure induced changes in  $\text{CuF}_2(\text{H}_2\text{O})_2(\text{pyz})$  as seen in x-ray diffraction (left) and magnetic data (right).

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[1] J. Manson *et al.*, Chem. Commun. (2006) 4894; [2] J. Manson *et al.*, Chem. Mater. 20 (2008) 7408.

[3] J. Manson *et al.*, J. Am. Chem. Soc. 131 (2009) 6733.