

Correlation between Cooperative Spin Crossover Phenomena and Intermolecular Interactions in Fe(III) Complexes

Tetsuya Sato¹, Kazuyuki Takahashi¹, Hatsumi Mori¹, Masaki Matsuda¹, Hiroyuki Tajima¹, and Osamu Sato²

¹*Institute for Solid State Physics, University of Tokyo, Japan*

²*Institute for Materials Chemistry and Engineering, Kyushu University, Japan*

Email: tetsuya@issp.u-tokyo.ac.jp

Spin crossover complexes show a spin state conversion induced by external stimuli such as temperature, pressure, and light. In the cooperative spin crossover phenomena, interactions between spin crossover molecules play an important role [1,2]. In order to investigate the relation between spin crossover phenomena and intermolecular interactions systematically, the preparations, crystal structure analyses, magnetic susceptibility measurements under pressures of $[\text{Fe}(\text{qsal})_2]\text{X}$ (Fig.1, X = linear anions), which are expected to possess strong π - π interactions in the crystal packing, were carried out.

$\text{Fe}(\text{qsal})_2$ complexes with linear anions were prepared by metathesis method. A newly prepared $[\text{Fe}(\text{qsal})_2]\text{I}_3$ is clarified to be the novel spin crossover complex with T_{SCO} (spin crossover temperature) \sim 250 K. The crystal structure analysis reveals that $\text{Fe}(\text{qsal})_2$ cations construct a one-dimensional chain with π - π interactions, and that the two-dimensional sheets composed of one-dimensional chains are separated by I_3 anions. As the pressure was applied, T_{SCO} shifts to higher temperatures by 7 K/kbar (Fig.2). On the other hand, $[\text{Fe}(\text{qsal})_2]\text{NCS}$ containing smaller SCN linear anion has the three-dimensional π - π interactions between one-dimensional chains and T_{SCO} shifts higher temperatures by 21 K/kbar with narrowing thermal hysteresis loop. These results indicate the correlation between cooperative spin crossover behaviors and dimensionality of intermolecular interactions in $[\text{Fe}(\text{qsal})_2]\text{X}$.

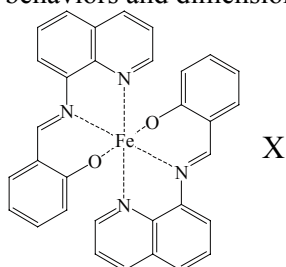


Fig.1 Molecular structure of $[\text{Fe}(\text{qsal})_2]\text{X}$

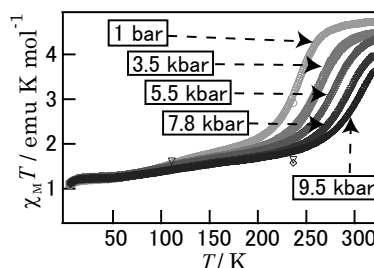


Fig.2 Pressure dependence of $\chi_M T$ for $[\text{Fe}(\text{qsal})_2]\text{I}_3$

[1] K. Takahashi *et al.*, *J. Am. Chem. Soc.*, 130 (2008) 6688.

[2] K. Takahashi *et al.*, *Polyhedron* (2009), doi:10.1016/j.poly.2008.11.047.